

Thermal expansion studies on As-Sb-Se glasses

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Abstract. The results of measurement of thermal expansion coefficient from 50°C to the softening temperature on eight glass compositions of the As-Sb-Se system are presented. Measurements have been made on $(As, Sb)_{40}Se_{60}$ and $As_xSb_{15}Se_{85-x}$ glasses at heating rates of 1°C/min and 5°C/min. The composition and temperature dependence of the thermal expansion coefficient is discussed in the light of results of other chalcogenide glasses.

Keywords. As-Sb-Se glasses; thermal expansion coefficient; glass transition region

1. Introduction

The study of thermal expansion coefficient (α) of chalcogenide glasses has not received much attention as compared with other properties of these glasses. There are some data on the composition dependence of α for As-Se (Webber and Savage 1976), As-S (Tsuchihashi and Kawamoto 1970), Ge-As-Se (Webber and Savage 1976) and Ge-Sb-Se (Sudha Mahadevan *et al* 1984) glasses. There is a scatter of about 15% in the reported value of α for As_2S_3 (Tsuchihashi and Kawamoto 1970; Hilton *et al* 1966) and of about 35% for As_2Se_3 (Henderson and Ast 1984; Webber and Savage 1976). It is generally known that the temperature dependence of α of glasses has two regions; an initial low temperature or 'normal' region where α is independent of temperature, followed by the glass transition region where α increases with increase of temperature. Measurement of α of Ge-Sb-Se glasses (Sudha Mahadevan *et al* 1984) using a conventional horizontal thermal dilatometer at a heating rate of 5°C/min indicated that the increase of α with temperature in the transition region occurred over a temperature range of about 30°C. This is followed by softening of the glasses. Values of α for As_2Se_3 measured using a parallel plate viscometer (Henderson and Ast 1984) indicate an abrupt increase of α (over a few degree temperature interval) from $15.5 \times 10^{-6}/^\circ C$ in the normal region to about $100 \times 10^{-6}/^\circ C$ after transition.

The results of measurement of α on eight glass compositions of the As-Sb-Se system are reported in this communication. The composition of four of these glasses can be represented by $(As, Sb)_{40}Se_{60}$. These fall along the $As_2Se_3 - Sb_2Se_3$ pseudo-binary tie line and constitute the so-called 'stoichiometric' glasses of the As-Sb-Se system. The composition of the other four glasses studied can be represented by $As_xSb_{15}Se_{85-x}$; these four along with the stoichiometric composition with 15 atomic percent of Sb, namely, $As_{25}Sb_{15}Se_{60}$ form five glasses of the $As_xSb_{15}Se_{85-x}$ family of the As-Sb-Se system. The results of measurement of the glass transition temperature (T_g), density, electrical conductivity and elastic properties of these glasses have been reported elsewhere (Giridhar and Sudha Mahadevan 1982; Giridhar *et al* 1984).

2. Experimental

The elemental components (5N purity, from Koch Light Co.) in appropriate atomic proportions (table I) were sealed in quartz ampoules (12mm diameter) under a vacuum of 10^{-3} torr. The contents of the ampoule were melted in a rotary furnace at about 950°C for 24 hr. The charge was then cooled to 800°C and quenched in cold water to obtain glasses. Samples were cut into rectangular sections of 3 mm thick \times 8 mm width and of length about 10 mm to 20 mm depending on the piece available in the glass preparation boule. The end faces normal to the length of the sample were rendered smooth and parallel using a hand polishing jig fabricated for this purpose. Before measurement each sample was annealed in situ by holding it for 30 min at a temperature of about 5°C higher than its T_g and cooling it slowly.

The expansion measurements were made in a conventional horizontal dilatometer (thermal dilatometric analyser, Harrop Laboratories, USA, model TDA-H1-PP2) which was standardised against an alumina rod supplied with the instrument. The sample temperature was measured using a calibrated chromel-alumel thermocouple in contact with the sample. Using an X-Y recorder (sensitivity 0.2 mV/cm., Watanabe model WX 431) in conjunction with the dilatometer, expansion of the samples were recorded from room temperature upto the respective softening temperature (T_s) at heating rates of 5°C/min and 1°C/min.

A typical trace showing the change in sample length Δl_s versus temperature is shown in figure 1 for one of the compositions. Also shown in figure 1 is the corresponding α calculated using the expression $\alpha = (l/l_s)(\Delta l_s/\Delta T)$, with a temperature increment of 10°.

3. Results and discussion

3.1 $(As, Sb)_{40}Se_{60}$ glasses

The results for these glasses are summarised in figures 2 and 3 and in table 1. Three to four batches of glasses were prepared for each composition to assess the spread in the value of α . While the samples from different batch preparations gave spread of about

Table 1. T_g , α and $\Delta\alpha$ data of As-Sb-Se glasses

Composition As Sb Se	T_g (°C) by DSC		Range of α (in the normal region) $\times 10^6$ (/°C)	$\Delta\alpha \times 10^6$ (/°C)	$V_f \times 10^6$ ($\approx 3\Delta\alpha T_g^*$)
	1.25°C /min	5°C /min			
10:15:75	86	92	24.0–234	65	0.072
20:15:65	135	142	21.0–23.0	59	0.073
30:15:55	175.5	183	17.5–18.8	55	0.075
35:15:50	179.5	187.5	18.0–20.7	50	0.069
40:0:60	164	171	19.4–20.7	70	0.093
35:5:60	167.5	175.5	18.6–21.0	53	0.071
30:10:60	171.5	178.5	18.3–20.5	60	0.081
25:15:60	174.5	181.5	17.7–20.0	55	0.075
22:18:60	175	182	17.2–19.0	54	0.074

* T_g in "K has been used

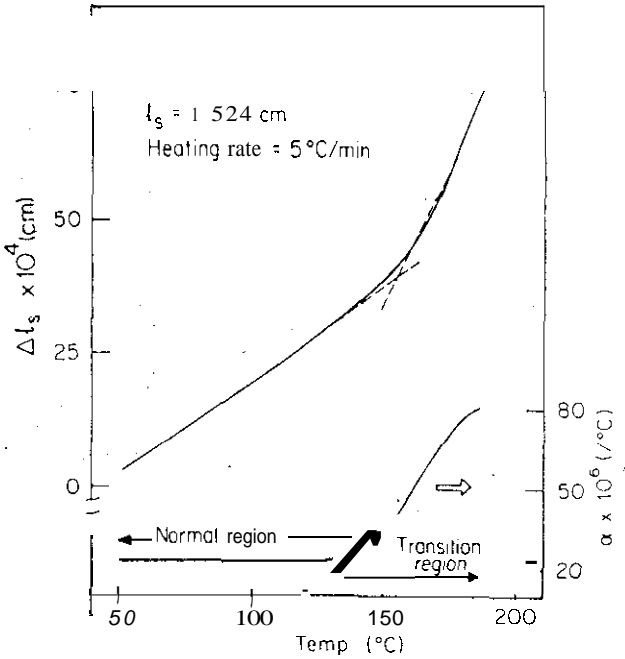


Figure 1. A typical data of the change in sample length Δl_s versus temperature for one of the compositions. Also shown is the corresponding α obtained by $\alpha = (l/l_s) (\Delta l_s/\Delta T)$.

6% to 12% depending on the composition, the value of α of samples from the same batch did not differ much. Because of the wide spread in the reported value of α for As_2Se_3 , two batches of $As_{40}Se_{60}$ glass were prepared and α was measured for the present study. The results for this composition are also included in figures 2 and 3. The T_g 's of these glasses obtained at heating rates of 1.25°C/min and 5°C/min using a Perkin Elmer model DSC-2 differential scanning calorimeter are also included in table 1. The temperatures at which α increases rapidly (figure 2) agree closely with T_g 's of these glasses.

The variation of α in the transition region of these glasses at the slow heating rate (1°C/min, dotted line, figure 2) indicates the existence of a relaxation process near T_g . It was found to be present in all the samples studied, though the magnitude of the effect differed. The origin of this relaxation process is not clear at present. It is probably similar to the relaxation process (Schnaüs *et al* 1970) normally encountered in the measurement of heat capacity (C_p) of glasses in the transition region. In heat capacity measurements it is attributed as an artifact of the rate dependence of glass properties in the transition region, and is associated with the structural relaxation times involved. However, further work is necessary to trace the origin of the relaxation process observed presently for these glasses.

The composition dependence of α of these glasses in the normal region is shown in figure 3 with the vertical bars indicating the range of α measured on samples from different batch preparations. A decrease of α with increase of Sb content in the glass is seen from the results of figure 3.

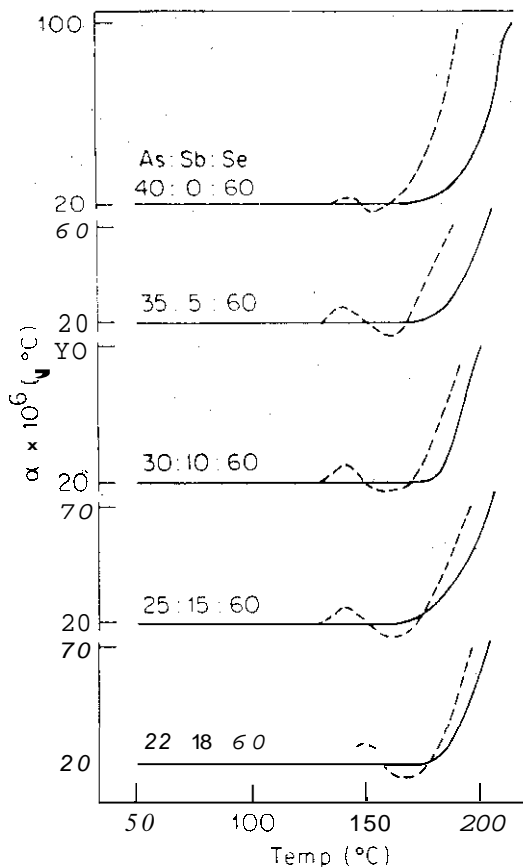


Figure 2. α -T data for five glasses of $(\text{As, Sb})_{40}\text{Se}_{60}$ (with Sb of 0, 5, 10, 15 and 18 atomic percent) at heating rates of $1^\circ\text{C}/\text{min}$ (broken lines) and $5^\circ\text{C}/\text{min}$ (continuous liner).

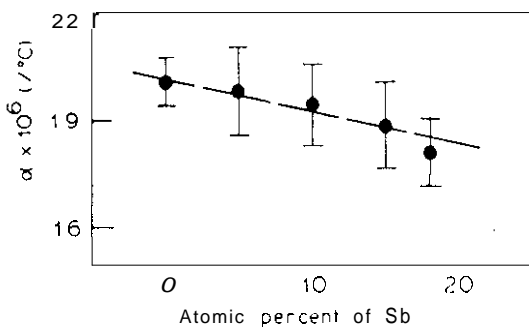


Figure 3. α in the normal region versus composition for $(\text{As, Sb})_{40}\text{Se}_{60}$ glasses. The vertical bars show the spread of α obtained by measurements on samples from different batch preparations.

3.2 $As_xSb_{15}Se_{85-x}$ glasses

The variation of α with temperature for glasses of this family was similar to those shown in figure 2 for stoichiometric compositions, with α being constant up to the corresponding T_g , and increasing rapidly beyond that. However, the relaxation effect was not as prominent as in the case of stoichiometric glasses. The variation with composition of α in the normal region for these glasses is shown in figure 4 along with the α -composition data of As_xSe_{100-x} , $Ge_xAs_{10}Se_{90-x}$ (Webber and Savage 1976), $Ge_xSb_{15}Se_{85-x}$ (Sudha Mahadevan *et al* 1984) and As_xS_{100-x} (Tsuchishshai and Kawamoto 1970) glasses. The lower values of α for the Ge rich glasses of the Ge-chalcogenide glasses (figure 4) can be understood on the basis that As-(S,Se) glasses are basically in the form of layers and chains while Ge-(S, Se) glasses are generally in the form of three-dimensional networks. In the composition dependence of α for Ge-Se (Feltz 1975), Ge-Sb-Se (figure 4) systems, a change of slope is seen at the respective stoichiometric composition, while for the As-S, As-Se systems (figure 4) a minimum is seen at the corresponding stoichiometric composition. A broad minimum is seen at the stoichiometric composition for the As-Sb-Se system. The variation in α with composition in $As_xSb_{15}Se_{85-x}$ is far less compared with those in case of As_xSe_{100-x} and As_xS_{100-x}

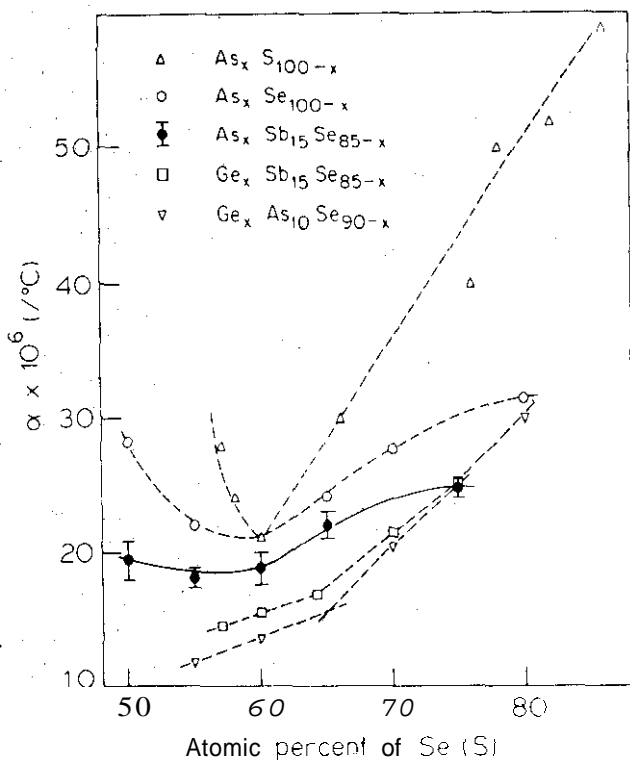


Figure 4. α -composition data for $As_xSb_{15}Se_{85-x}$ glasses. Also shown in the graph are the α - T data from the literature for As_xS_{100-x} (Tsuchihashi and Kawamoto 1970), As_xSe_{100-x} , $Ge_xAs_{10}Se_{90-x}$ (Webber and Savage 1976) and $Ge_xSb_{15}Se_{85-x}$ (Sudha Mahadevan *et al* 1984) glasses.

(figure 4) glasses. The addition of Sb seems to suppress the variation of α with composition.

3.3 Variation of α in the transition region

In the normal region, the α of these glasses is not dependent on the temperature or the heating rate employed for the measurement. In the transition region, a marked dependence of α is seen both on the temperature and on the heating rate. This behaviour is similar to the behaviour of heat capacity of glasses in the transition region (Schnaus *et al* 1973). The region of rapidly increasing α is shifted to higher temperature as the heating rate is increased (figure 2), a behaviour similar to that of the heat capacity of glasses.

These measurements, as many others made earlier, indicate that α increases in the transition region. The present measurements have been made with a dilatometer wherein a probe rod under load makes contact with the sample. The temperature in the set-up has been varied continuously at a chosen heating rate. The results for As_2Se_3 at heating rates of $1^\circ\text{C}/\text{min}$ and $5^\circ\text{C}/\text{min}$ are shown in figure 5 (traces a and b respectively). In these cases, α has been calculated from the slope of the Δl_s versus T trace (similar to that of figure 1) with a 10°C temperature increment. The increase in α is spread over a temperature of about 45°C for $5^\circ\text{C}/\text{min}$ heating rate, and about 30°C for $1.25^\circ\text{C}/\text{min}$ heating rate. It is interesting to note in this context that the expansion data of As_2Se_3 and As_2S_3 -PbS glasses (Bhat and Bhatia 1978) also indicate an increase in α taking place over a temperature range of about 40°C in the transition region. Recent results (Ahmed *et al* 1985) on thermal expansion of mixed alkali borate glasses also indicate an increase in α occurring over a temperature interval of more than 75°C around the transition region for some of the compositions. This is to be expected because the transformation to a glass does not take place at one, strictly defined temperature, but within a temperature range, representing the transformation region (Maurer 1968; Vacher *et al* 1974; Rehage and Rorchard 1973). However, it is a general

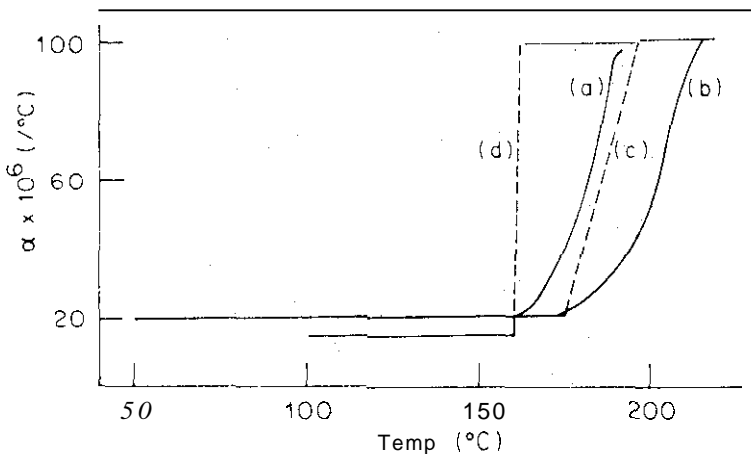


Figure 5. α - T in the transition region for As_2Se_3 at the heating rates of $1^\circ\text{C}/\text{min}$ (trace a) and $5^\circ\text{C}/\text{min}$ (trace b). Trace c is calculated from the data obtained at the heating rate of $5^\circ\text{C}/\text{min}$ by approximating the Δl_s vs T in the transition region to a straight line. The data of Henderson and Ast (1984) obtained from viscometer measurement is shown in trace d.

practice to approximate the variation of Δl_s with T in the transition region to a straight line (as shown by the dotted line in figure 1) and ascribe a value for a after transition as inferred from the slope of this line. The result of such a calculation (for the heating rate of $5^\circ\text{C}/\text{min}$) for As_2Se_3 is as shown in trace (c) of figure 5. Also shown in figure 5 are the data on a of As_2Se_3 obtained by Henderson and Ast (1984) (trace (d), figure 5). The maximum value of a of about $100 \times 10^{-6}/^\circ\text{C}$ obtained presently agrees well with that obtained by Henderson and Ast (1984). However, the manner in which a increases in the transition region before reaching its maximum value is different in the two measurements. In the set-up used by Henderson and Ast, the sample is placed between the plates of a parallel plate viscometer and the temperature is increased in steps. In this sense, these measurements are under isothermal conditions. Henderson and Ast (1984) also reported that the load used did not cause any deformation of the sample. The α in this case is seen to increase to its final value in a few degrees range of temperature, whereas the present measurements indicate that the increase is spread over a wider range of temperature. This feature is introduced because of the continuous heating. The increase in a is spread over a range of 45°C for $5^\circ\text{C}/\text{min}$ heating rate and about 30°C for the $1.25^\circ\text{C}/\text{min}$ heating rate. Also, by approximating the Δl_s vs T in the transition to a straight line, the spread of temperature range over which the increase in a occurs is narrowed down, as can be seen from trace (c) of figure 5. It is to be expected that at still slower heating rates (approaching isothermal conditions) the spread will further decrease and resemble the data of Henderson and Ast (1984). Isothermal measurements were attempted on the dilatometer to check this point. But this could not be checked because the probe rod started indenting the sample when the sample was held for a sufficiently long time at any temperature in the transition region, indicating an apparent decrease in length. Thus while there is an increase of a in the transition region, the detailed nature of this increase seems to depend on the heating rate employed, the procedure used to evaluate a (from the Δl_s vs T data) and the method employed for the measurement of a .

The free volume V_f during glass transition is given (Simha and Boyer 1962; Suzuki *et al* 1971) as the product of the difference in cubical expansion coefficient above and below the glass transition temperature and $T_g \cdot V_f$ is of the order of $3Aa T_g$, where Aa is the change in linear expansion coefficient above and below T_g . For oxide glasses, the ratio of free volumes, namely, $V_{f(\text{B}_2\text{O}_3)}/V_{f(\text{GeO}_2)}$ is found to be equal to eight, while by considering the average free volume changes in As-Se, As-S, Ge-Se and Ge-S systems, it has been found that $V_{f(\text{As}_2\text{Se}_3)}/V_{f(\text{GeSe}_2)}$ is about 1.2 and $V_{f(\text{As}_2\text{S}_3)}/V_{f(\text{GeS}_2)}$ is about 1 (Suzuki and Abe 1981). It is therefore concluded (Suzuki and Abe 1981; Suzuki *et al* 1978) that chalcogenide glasses are in a state of isofree volume during glass transition. The average value of V_f for (As_2Se_3) (Sb_2Se_3) glasses (table 1) is seen to be 0.0789. The average value of V_f from the data of Aa and T_g for (GeSe_2) (Sb_2Se_3) glasses (Sudha Mahadevan *et al* 1984) is seen to be 0.0760. The value of 1.04 for the ratio $V_{f(\text{As}_2\text{Se}_3)/(\text{Sb}_2\text{Se}_3)}/V_{f(\text{GeSe}_2)/(\text{Sb}_2\text{Se}_3)}$ therefore falls in line with the general observation of isofree volume during glass transition for chalcogenide glasses, despite the difference in structure.

4. Summary and conclusions

Measurements have been made on the dependence of thermal expansion coefficient (a) on temperature and composition for eight glass compositions of the As-Sb-Se system at

the heating rates of $1^{\circ}\text{C}/\text{min}$ and $5^{\circ}\text{C}/\text{min}$. The a - T data indicate an initial 'normal' region where a is independent of temperature followed by a 'transition' region where a increases steeply with temperature. The temperature where a begins to show a steep increase shifts to higher values as the heating rate is increased. The detailed nature of the increase of a in the transition region seems to depend on the heating rate employed, the procedure used to evaluate a and the method employed for the measurement of a .

The composition dependence of a of $\text{As}_x\text{Sb}_{15}\text{Se}_{85-x}$ glasses have been compared with those of $\text{As}_x\text{S}_{100-x}$ and $\text{As}_x\text{Se}_{100-x}$ glasses. A broad minimum is seen at the 'stoichiometric' composition $\text{As}_{25}\text{Sb}_{15}\text{Se}_{60}$ in the a -composition dependence of these glasses.

The a in the 'normal' region for $(\text{As}, \text{Sb})_{40}\text{Se}_{60}$ glasses shows a slight decrease with increasing Sb content in the glasses. At the heating rate of $1^{\circ}\text{C}/\text{min}$, these glasses exhibit a relaxation process around the transition region. The results on a and T_g of these glasses along with those of Ge-Sb-Se glasses fall in line with the general observation of iso-free volume during glass transition for chalcogenide glasses, despite differences in structure.

References

- Ahmed A A, Abbas A F and Salman S M 1985 *Phys. Chrm. Glasses* **16** 17
 Bhat P K and Bhatia K L 1978 *Phys. Chem. Glasses* **19** 55
 Feltz A 1975 *Z. Anorg. Allg. Chem.* **412** 20
 Giridhar A and Sudha Mahadevan 1982 *J. Non-Cryst. Solids* **51** 305
 Giridhar A, Sudha Mahadevan and Singh A K 1984 *Bull. Mater. Sci.* **6** 1001
 Henderson W and Ast D G 1984 *J. Non-Cryst. Solids* **64** 43
 Hilton A R, Jones C E and Brau M 1966 *Phys. Chem. Glasses* **1** 105
 Kawamoto Y and Tsuchihashi S 1971 *J. Am. Cer. Soc.* **54** 131
 Maurer J J 1968 in *Analytical chemistry VI* (New York: Plenum Press) pp 107
 Rehage C and Borchard W 1973 in *The physics of glassy polymers* (London: Applied Science Publications) PP 54
 Schnaus V E, Moynihan C T, Gammon R W and Macedo P B 1970 *Phys. Chem. Glasses* **11** 213
 Simha R and Boyer R F 1962 *J. Chem. Phys.* **37** 1003
 Sudha Mahadevan, Giridhar A, Narasimham P S and Singh A K 1984 *J. Non-Cryst. Solids* **65** 233
 Suzuki S and Abe Y 1981 *J. Non-Cryst. Solids* **43** 141
 Suzuki S, Takahashi M and Kobayoshi T 1978 *Yogyo Kyokai Shi* **86** 428
 Tsuchihashi S and Kawamoto S 1970 *J. Non-Cryst. Solids* **5** 286
 Vacher R, Delsanti M, Pelous J, Cecchi L, Winter A and Zarzycki J 1984 *J. Mater. Sci.* **9** 829
 Webber P J and Savage J A 1976 *J. Non-Cryst. Solids* **20** 271